

Retrieval of atmospheric sulfur dioxide and nitric acid using the Atmospheric Infrared Sounder (AIRS)

Scott Hannon and L Larrabee Strow

Physics Department and Joint Center for Earth Systems Technology, University of Maryland
Baltimore County, 1000 Hilltop Circle, Baltimore, MD 21250

ABSTRACT

The Atmospheric Infrared Sounder (AIRS) flying on NASA's EOS-AQUA platform has channels sensitive to both sulfur dioxide (SO₂) and nitric acid (HNO₃). We have developed a simple regression retrieval for both of these gases that illustrates the potential for AIRS, and other hyperspectral sounders, to retrieve these two gases. We have cross-validated the SO₂ retrievals to those from the OMI instrument flying on the EOS AURA platform. Similarly, we have cross-validated the HNO₃ retrievals with limb retrievals of HNO₃ by from MLS instrument, also flying on the AURA platform.

Keywords: remote sensing, minor gases, sulfur dioxide, nitric acid, atmospheric retrievals

1. INTRODUCTION

The Atmospheric Infrared Sounder (AIRS) is a high spectral resolution grating spectrometer on board the NASA's EOS Aqua satellite platform, and has been in near continuous operation since September 2002. The sun-synchronous orbit of Aqua combined with the nadir viewing cross-track scanning of AIRS provides observations of the majority of Earth's surface twice each day, once during the day and again ~ 12 hours later at night. AIRS has 2378 spectral channels spanning 650-1615 and 2180-2670 cm^{-1} with a resolution of approximately $\nu/1200$. A number of these channels are sensitive to the atmospheric concentration of trace gases SO₂ (sulfur dioxide) and HNO₃ (nitric acid). See a recent review article on AIRS¹ and references therein for an overview of the AIRS instrument and retrieval products.

The "version 5" AIRS operational retrieval software package used by the Goddard Earth Sciences Data Information Services Center (GES DISC) to process AIRS observations and produce data products (temperature, water vapor, ozone) does not include retrievals of either SO₂ or HNO₃. This paper gives an overview of some preliminary work to explore the potential of AIRS to retrieve these trace gases. A relatively simple retrieval technique is used, with encouraging results.

2. SULFUR DIOXIDE

Under ordinary conditions, SO₂ is present in the atmosphere at background levels on the order of 0.1 Dobson Unit (DU). A Dobson Unit is the thickness in hundredth of a millimeter of a gas compressed to a pressure of 1 atmosphere at 0 C. This low background level is too low for AIRS to detect, but volcanic eruptions often release large amounts of SO₂ into the atmosphere creating SO₂ plumes of tens or hundreds of DU that can travel long distances before diffusing away. AIRS has some ability to detect and measure these SO₂ plumes when the gas concentration is well above typical background levels. We have previously detected SO₂ with AIRS from a major eruption of Mt. Etna,² using a preliminary retrieval algorithm. In that work we also showed that AIRS can detect and track the evolution of a volcanic ash cloud if it exists with sufficient optical depth. Note that the spectral signatures of the ash and SO₂ clouds do not overlap, allowing both to be retrieved without interference from the other.

AIRS channels with some sensitivity to the concentration of SO₂ present in the observation are shown in the top panel of Figure 1, which shows a simulated AIRS clear sky brightness temperature (BT) spectra for the 950-1450 cm^{-1} region for three SO₂ profiles of 0.1, 100, and 1000 DU. The bottom panel shows the change in BT due to the change in SO₂. Each year there are usually a few eruptions somewhere on the globe with SO₂ plumes on the order of 100 DU, while eruptions on the order of 1000 DU are very rare (Pinatubo in 1991 was the most recent).

As can be seen from Fig. 1, the AIRS channels most sensitive to changing SO₂ are located in the 1370 cm^{-1} region. For a typical background level SO₂ profile, the atmospheric absorption there is dominated by water vapor, but also with significant contributions by CH₄ and N₂O. With a careful selection of channels the CH₄ and N₂O can be avoided, but the water vapor affects all the 1370 cm^{-1} region channels. The water vapor absorption here blocks most of the upwelling

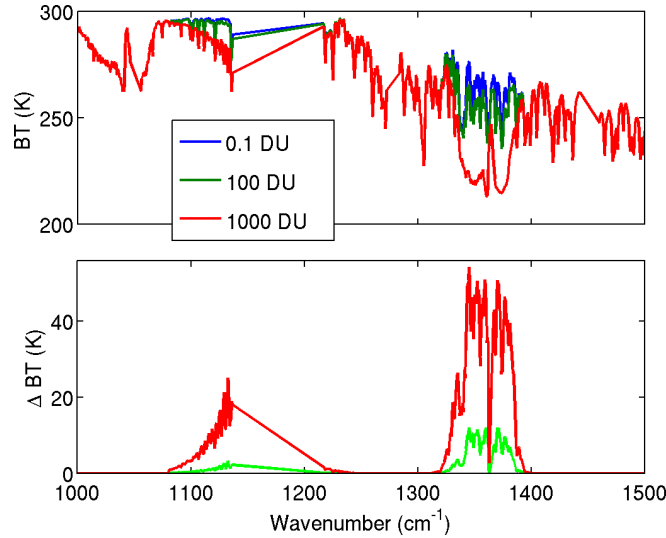


Figure 1. Sensitivity of AIRS to SO₂.

radiance originating at the surface and the lower troposphere, thereby preventing these channels from detecting low altitude SO₂ plumes. The 1100 cm⁻¹ region channels can see to the surface, but they can only detect SO₂ if it is on the order of 100 DU or more.

Our SO₂ (and HNO₃) retrieval method is based on regression. We started by selecting two sets of AIRS channels: one set with strong SO₂ absorption, and another set with weak SO₂ but with otherwise similar absorption characteristics as the strong SO₂ channels. That is, we chose our channels such that with nominal background level SO₂ the two sets of channels respond to the atmospheric state in similar ways and AIRS observes similar BTs. With such a channel set it may be possible to use the BT of "weak" channels to predict the BT of the "strong" channels under the assumption of nominal background level SO₂. We do a simple regression of the form

$$A \times X = B \quad (1)$$

where A is a matrix of "weak" channel observed BTs (the predictor matrix), X are the coefficients to be determined by regression, and B are the "strong" channel BTs. The regression was trained on simulated spectra (including random noise), and once the coefficients X are determined the above equation can be used to predict the "strong" BT using real observed data.

If the actual SO₂ present in the observation is substantially higher than the nominal level, then the BT predicted for the "strong" channels will differ from the actual measured BT. The size of this ΔBT will depend on the amount of excess SO₂. Our AIRS fast model^{3,4} allows for variable SO₂, so if we have an estimate of the atmospheric temperature and water vapor profiles, we can compute the expected delta BT for a given amount of SO₂. Using an iterative approach we can then estimate the amount of SO₂ present in the observation.

Figure 2 shows the AIRS channels we selected for use in our SO₂ retrieval. The top panel shows a blowup of the BT spectra, while the bottom panel shows AIRS sensitivity to changes in SO₂ as well as CH₄ and N₂O. The open circles are channels sensitive to SO₂, while the solid dots are the SO₂ insensitive channels. Note the avoidance of channels strongly sensitive to CH₄ and N₂O in the bottom panel. The effect of increasing the SO₂ from 0.1 to 1 DU shown in the bottom panel changes the AIRS BT by at most 0.15 K, which is similar to the AIRS noise level. Thus 1 DU is a rough estimate of the lower limit of AIRS sensitivity to SO₂.

Another consideration for AIRS SO₂ retrievals are cloud effects. Besides the usual cloud cover found across the globe, volcanic eruptions releasing large amounts of SO₂ also tend to release substantial amounts of water vapor and sometimes also ash. As the volcanic steam rises, it will cool and form clouds. These volcanic clouds often reach high into the troposphere, sometimes all the way to the tropopause. AIRS will not be able to detect the SO₂ below these clouds.

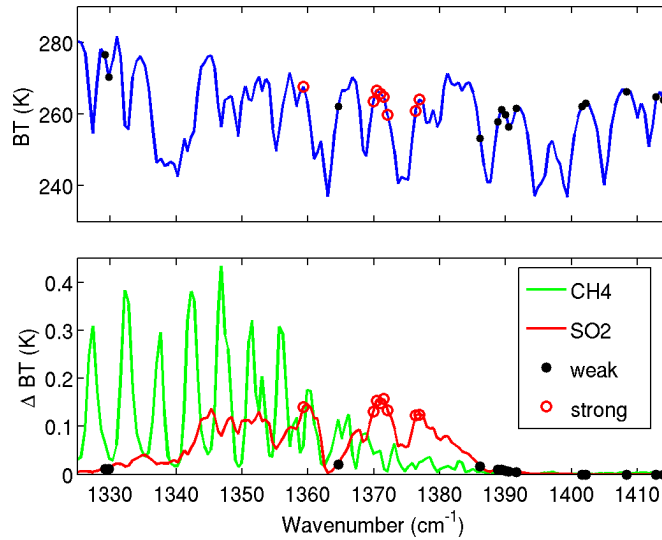


Figure 2. AIRS channels selected for SO₂ retrievals. Top panel is an AIRS simulated spectrum, the bottom panel shows the sensitivity of this spectrum to SO₂, CH₄, and N₂O.

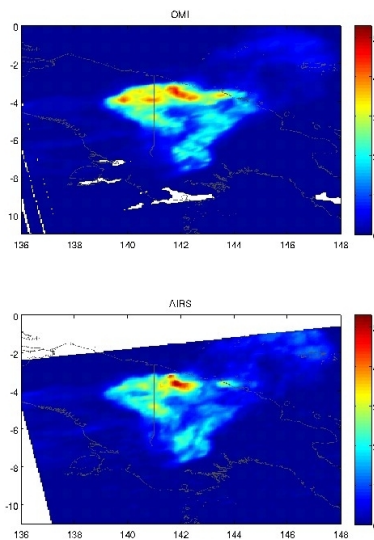


Figure 3. Comparison of SO₂ cloud distribution and amount between OMI and AIRS. Colorbar units are Dobson units of SO₂. Map scales are latitude (horizontal) and longitude (vertical) in degrees.

AIRS has very limited ability to determine the vertical location of an SO₂ plume. If we can clearly see an SO₂ plume using the 1370 cm⁻¹ channels then we know a significant amount of SO₂ must have reached the upper troposphere, but we do not know the shape of the profile. That is, we can not tell if the SO₂ plume extends all the way from the tropopause down to the surface, or if it is restricted to some sort of slab in the upper troposphere. In general, AIRS is only sensing a portion of the vertical column, and so we must assume a vertical profile shape when estimating the total column.

Figure 3 shows our AIRS retrieval of an SO₂ plume near Manam in New Guinea on 28 January 2005. Also shown for comparison is the SO₂ retrieval from the Ozone Monitoring Instrument (OMI).⁵ For this comparison the AIRS retrieval assumed the SO₂ resided in a slab confined to approximately 200-350 mb. On the whole, the retrieved SO₂ agrees well with OMI, but AIRS SO₂ column values are dependent on the assumed vertical location of the SO₂ plume. Moving the assumed location 100-200 mb gives a retrieval that looks almost exactly the same except the column amounts are halved!

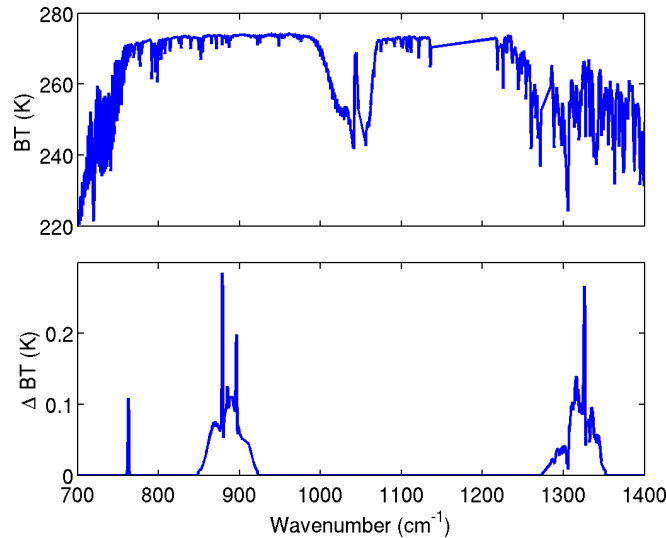


Figure 4. Sensitivity of AIRS to HNO₃. Top panel is a simulated AIRS spectrum. The bottom panel is the change in BT for a 50% increase in the HNO₃ column.

In some case, clouds can be used to help provide information about the vertical location of the SO₂ plume. If the field of view of an AIRS observation is completely filled by a cloud, then AIRS channels that ordinarily see to the surface in clear sky conditions will instead see a BT corresponding to the temperature at the top of the cloud. Combined with knowledge of temperature profile, this can be used to estimate the vertical location of cloud top. If AIRS observes enhanced SO₂ under these conditions, then we know the SO₂ plume extends above the cloud top. The tropopause tends to act as a cap on mixing of air from the tropopause and stratosphere, so we can also assume the SO₂ plume will rarely extend above the tropopause.

3. NITRIC ACID

Our retrieval method for HNO₃ is similar in concept to that discussed for SO₂ above, with some important differences. Unlike SO₂ which is subject to huge variations in concentration due to volcanos, the range of variation of HNO₃ is much smaller, and we want to measure HNO₃ amounts less the nominal mean as well values greater. Also most of the variations in HNO₃ column are found in the stratosphere and not the troposphere as was the case for SO₂. The nominal global average of HNO₃ is about 0.4 DU, and the range of variation is roughly 0.2 to 5 times the nominal mean.

The top panel of Fig. 4 shows a typical AIRS spectrum for the 650-1620 cm⁻¹ channels, and the bottom panel shows the delta BT resulting from increasing HNO₃ by 50%. There are two clumps of AIRS channels with strong sensitivity to HNO₃: one in the 880 cm⁻¹ window region, and the other in the 1330 cm⁻¹ region.

The 1330 cm⁻¹ channels are dominated by water absorption, but there is also significant CH₄ and N₂O absorption in some channels which our retrieval avoided. The water absorption blocks most radiation from the surface and lower troposphere for typical tropical and mid-latitude profiles, but over the poles where the air is usually very dry and the surface altitude is often high, these lower troposphere/surface radiance contributions can become non-negligible.

The 880 cm⁻¹ channels are a little noisier than 1330 cm⁻¹, but much less affected by other absorbing gases. Surface and lower tropospheric radiance contributions dominate total radiance observed in the 880 cm⁻¹ channels, so this makes the channels more sensitive to surface modeling errors and the effects of low altitude clouds than the 1330 cm⁻¹ channels. For these reasons we used the 1330 cm⁻¹ channels for our retrieval.

Due to the relative small range in delta BT due to HNO₃ variability, our retrieval requires an accurate atmospheric temperature and water vapor profile, and cloud effects need to be accounted for. We therefore chose to do our HNO₃ retrieval using the AIRS Level 2 support products profile and cloud clear radiances.

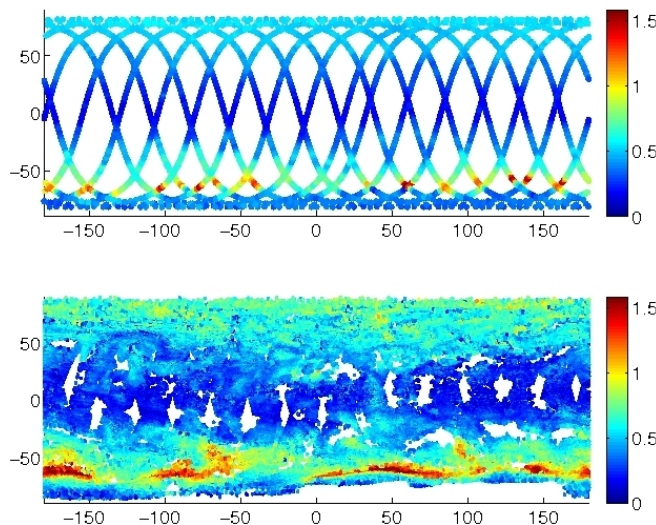


Figure 5. Comparison between observed HNO₃ from MLS (top panel) and AIRS (bottom panel). Colorbar units are Dobsons. Map scales are latitude (horizontal) and longitude (vertical) in degrees.

Figure 5 shows a comparison of our global AIRS HNO₃ retrieval for 22 July 2006 and the version 1.5 HNO₃ product of the Microwave Limb Sounder (MLS).⁶ The spatial coverage of MLS is much less than AIRS, but MLS HNO₃ should be more accurate than AIRS. The agreement of the two retrievals for this particular case is fairly good, but in other cases we see offsets between the two instruments. AIRS retrievals tend to be biased high compared to MLS, and AIRS tends to have more HNO₃ over the poles than MLS. In addition there is a small but noticeable limb-darkening effect bleeding into the AIRS retrievals. However, these results are quite encouraging given the simple approaches we have used to date.

4. CONCLUSIONS

We have demonstrated that AIRS has some ability to retrieve quantitative information about volcanic SO₂ plumes and global HNO₃.

Acknowledgments

We would like to thank Sergio De-Souza Machado and Howard Motteler at UMBC for their contributions to this work.

REFERENCES

1. M. T. Chahine, T. S. Pagano, H. H. Aumann, R. Atlas, C. Barnet, J. Blaisdell, L. Chen, M. Divakarla, E. J. Fetzer, M. Goldberg, C. Gautier, S. Granger, S. Hannon, F. W. Irion, R. Kakar, E. Kalnay, B. H. Lambrigtsen, S.-Y. Lee, J. Le Marshall, W. W. McMillan, L. McMillin, E. T. Olsen, H. Revercomb, P. Rosenkranz, W. L. Smith, D. Staelin, L. Larrabee Strow, J. Susskind, D. Tobin, W. Wolf, and L. Zhou, "AIRS: Improving Weather Forecasting and Providing New Data on Greenhouse Gases.," *Bulletin of the American Meteorological Society*, vol. 87, Issue 7, pp.911-926 **87**, pp. 911–926, July 2006.
2. S. A. Carn, L. L. Strow, S. de Souza-Machado, Y. Edmonds, and S. Hannon, "Quantifying tropospheric volcanic emissions with AIRS: The 2002 eruption of Mt. Etna (Italy)," *Geophys. Res. Lett.* **32**, pp. 2301–+, Jan. 2005.
3. L. L. Strow, S. E. Hannon, S. De-Souza Machado, H. E. Motteler, and D. C. Tobin, "Validation of the Atmospheric Infrared Sounder radiative transfer algorithm," *Journal of Geophysical Research (Atmospheres)* **111**, pp. 9–+, May 2006.
4. L. Strow, S. Hannon, S. DeSouza-Machado, and H. Motteler, "An overview of the AIRS radiative transfer model," *IEEE Transactions on Geosciences* **41**(2), pp. 303–313, 2003.
5. K. Yang, N. A. Krotkov, S. A. Carn, A. J. Krueger, and P. F. Levelt, "A Fast Algorithm for Retrieval of Volcanic Sulfur Dioxide With OMI," *AGU Fall Meeting Abstracts*, pp. A1737+, Dec. 2006.

6. M. L. Santee, N. J. Livesey, Q. Li, G. L. Manney, A. Lambert, and W. G. Read, "Upper Tropospheric Nitric Acid from the Microwave Limb Sounder on Aura: First Results," *AGU Fall Meeting Abstracts* , pp. B848+, Dec. 2006.